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Publication details, including instructions for authors and subscription information:

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Online publication date: 12 August 2010

**To cite this Article** Zak, Zdirad and Taraba, Jan(2004) 'NEW POLYHETEROATOMIC CYCLIC MOLECULES CONTAINING ELEMENTS of the 13.-16. GROUP', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 179: 4, 839 — 843

**To link to this Article:** DOI: 10.1080/10426500490427286

**URL:** <http://dx.doi.org/10.1080/10426500490427286>

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## NEW POLYHETEROATOMIC CYCLIC MOLECULES CONTAINING ELEMENTS OF THE 13.–16. GROUP

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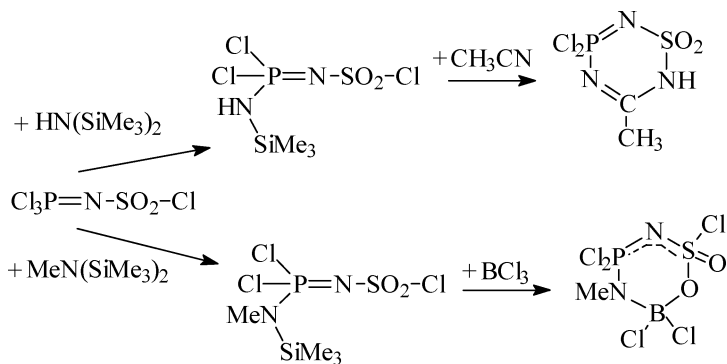
(Received August 17, 2003; accepted October 3, 2003)

*During studies of the reactions of  $-N(H)SiMe_3$  and  $-N(Me)SiMe_3$  derivatives of  $Cl_3PNSO_2Cl$  with acetonitrile and  $BCl_3$  we have obtained six-membered polyheteroatomic cycles  $[P(Cl_2)NSO_2(Cl)N(H)C(Me)N]$  and  $[P(Cl_2)NS(O)(Cl)OB(Cl_2)N(Me)]$ .<sup>1,2</sup> In the system  $Ph_2PCl_3$ ,  $H_2NSO_2Cl$  and  $HN(SiMe_3)_2$  we have identified and isolated several P–N–S cycles, e.g. the reaction of  $Ph_2PCl_3$  with  $H_2NSO_2Cl$  gives  $Ph_2ClPNSO_2Cl^3$  which with  $HN(SiMe_3)_2$  reacts to  $[S(O_2)N(H)P(Ph)_2N(H)SO_2N(H)P(Ph)_2N(H)]$ ,  $[S(O_2)N(H)S(O_2)N(H)P(Ph)_2N(H)P(Ph)_2N(H)]$  and  $[S(O_2)N(H)P(Ph)_2NP(Ph)_2N(H)]^+ Cl^-$ ;  $Ph_2PCl_3$  with  $HN(SiMe_3)_2$  gives  $N[P(Ph)_2N(H)SiMe_3]_2^+ Cl^-$ , and  $H_2NSO_2Cl$  with  $HN(SiMe_3)_2$  leads to  $SO_2(NHSiMe_3)_2$ . The reaction of  $Ph_2PCl_3$  with  $HN(SiMe_3)_2$  gives  $N(P(Ph)_2NHSiMe_3)_2Cl$  in a very good yield which was further used to syntheses of metal-containing heterocycles. By the reaction of  $N[P(Ph)_2N(H)SiMe_3]_2^+ Cl^-$  with some covalent halogenides we have obtained six-membered heterocycles containing B, As, In, and Sn. The same cyclic compounds can also obtained by the reaction of  $N[P(Ph)_2NH_2]_2^+ Cl^-$  or  $HN(P(R)_2N(H)SiMe_3)_2$  with covalent halogenides.<sup>4–6</sup> However, the synthetic route via  $N[P(Ph)_2NHSiMe_3]_2^+ Cl^-$  is more convenient and gives the compounds in almost quantitative yields. The identity of all compounds was unambiguously established by their X-ray structure determination.*

**Keyword:** Heterocyclophosphazenes

During our previous studies of the reactions of  $-N(H)SiMe_3$  and  $-N(Me)SiMe_3$  derivatives of  $Cl_3PNSO_2Cl$  (1) with acetonitrile and  $BCl_3$ , we have obtained and structurally fully characterized two new six-member polyheteroatomic cycles according to the Scheme 1.<sup>1,2</sup>

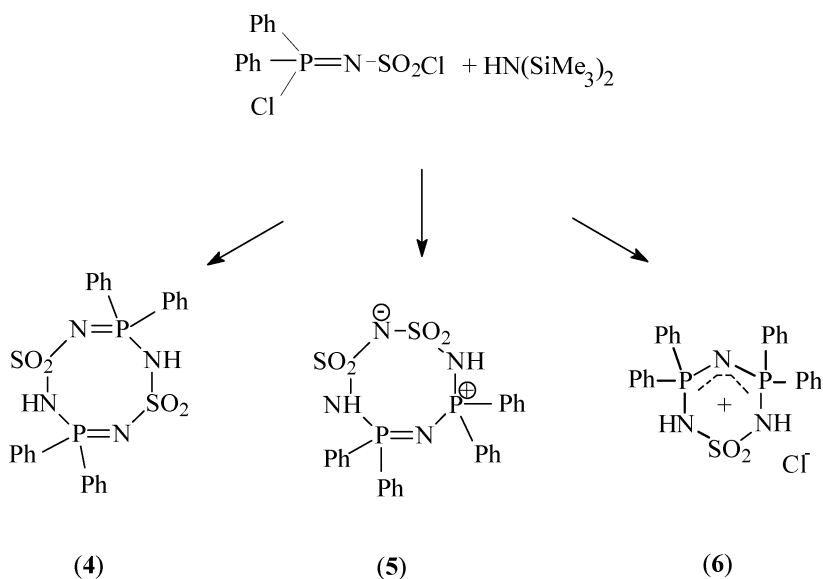
Address correspondence to Zdirad Zak, Department of Inorganic Chemistry, Masaryk University, Brno, Czech Republic. E-mail: zak@chemi.muni.cz



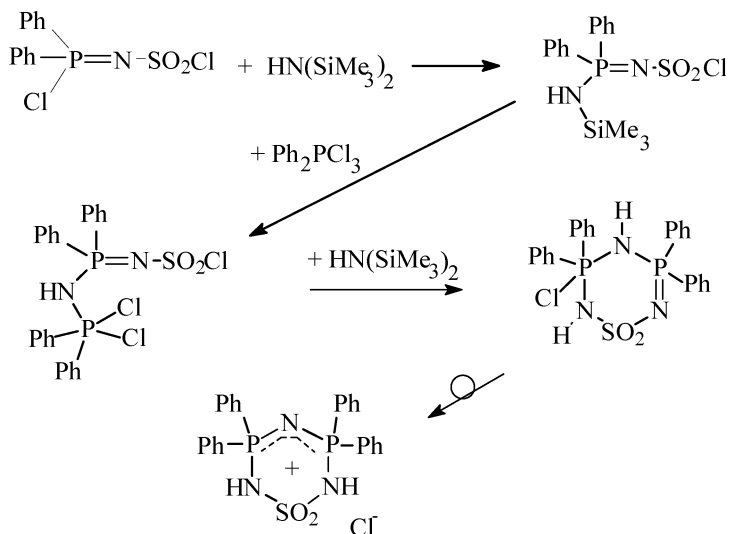
**SCHEME 1** Formation of P–N–C and P–N–S–O–B six-member cycles.

In the present research we have concentrated on the reaction of diphenyl derivative of **1**,  $\text{Ph}_2\text{ClPNSO}_2\text{Cl}$  (**2**) and a phosphorus analogue of **1**,  $\text{Cl}_3\text{PNPOCl}_2$  (**3**) with  $\text{HN}(\text{SiMe}_3)_2$  (HMDS) and  $\text{MeN}(\text{SiMe}_3)_2$  (HpMDS), respectively, and their subsequent reactions with some covalent halogenides.

From the reaction mixture of **2** with HMDS we have isolated three P–N–S cycles, two eight-member neutral molecules and one six-member cation isolated as a chloride salt (Scheme 2).

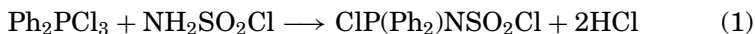


**SCHEME 2** Products from the reaction system  $\text{ClPPh}_2\text{NSO}_2\text{Cl}$  with HMDS.



**SCHEME 3** Reaction mechanism of the formation of P-N-S six-member cation in the system  $\text{ClPPh}_2\text{NSO}_2\text{Cl}$  with HMDS.

The compound **2** is prepared according to Haubold and Fluck.<sup>3</sup>



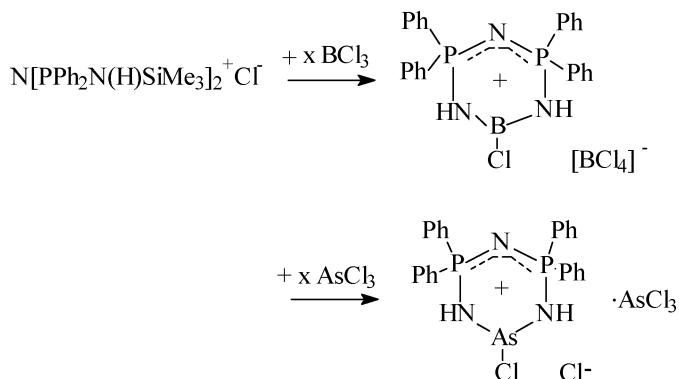
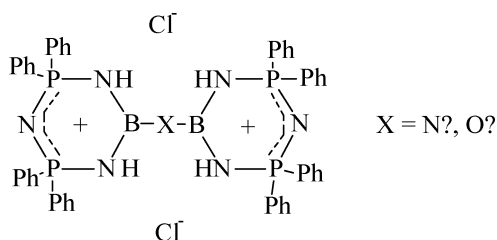
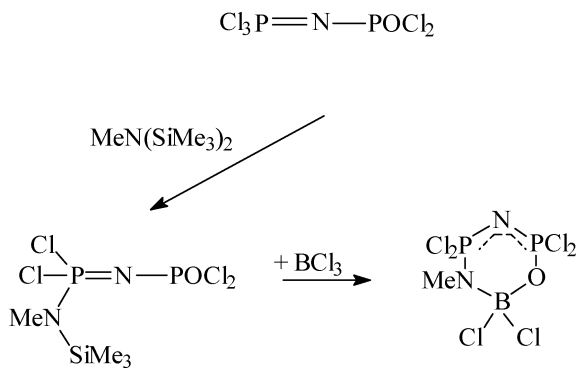
While the formation of eight-member cycles **4** and **5** can be explained easily by a subsequent condensation of, e.g., one molecule of **2** substituted by  $-\text{N}(\text{H})\text{SiMe}_3$  group both on P and S atoms with an unsilylated one (head-to-tail or head-to-head), the formation of **6** is more complex and requires, e.g., a presence of unreacted phosphorane in the reaction mixture (Scheme 3).

However, the reaction 1 is in fact an equilibrium reaction with approx. 80% conversion and thus the reaction mixture contains not only product **2** but the reactants, too. We have therefore separately examined reactions of the reactants,  $\text{Ph}_2\text{PCl}_3$  and  $\text{H}_2\text{NSO}_2\text{Cl}$ , with HMDS.

$\text{H}_2\text{NSO}_2\text{Cl}$  reacts with HMDS to give  $\text{N},\text{N}'\text{-bis}(\text{trimethylsilyl})\text{-sulfamide}$   $\text{SO}_2[\text{N}(\text{H})\text{SiMe}_3]_2$  with almost quantitative yield, while the reaction of  $\text{Ph}_2\text{PCl}_3$  yields acyclic phosphazanium salt  $[\text{Me}_3\text{SiN}(\text{H})\text{PPh}_2\text{NPPH}_2\text{N}(\text{H})\text{SiMe}_3]^+\text{Cl}^-$  (**7**).

This salt **7** was subsequently used to the synthesis of boron and arsenic containing six-member heterocyclic cations (Scheme 4).

If the reaction of **7** with  $\text{BCl}_3$  was performed in 1:1 molar ratio, we have obtained by a still not quit clear reaction mechanism a dimeric planar cyclic cation (Scheme 5).

**SCHEME 4** Reaction of **7** with  $\text{BCl}_3$  and  $\text{AsCl}_3$ .**SCHEME 5** Dimeric P–N–B cation.**SCHEME 6** Formation of P–N–O–B six-member cycle.

$\text{Cl}_3\text{PNPOCl}_2$  (**3**) reacts with HpMDS forming  $\text{Me}_3\text{SiN}(\text{Me})\text{PCl}_2\text{NPOCl}_2$  which gives with  $\text{BCl}_3$  a cyclic P–N–O–B compound (Scheme 6).

The identity of all compounds, which have been described was unambiguously established by their x-ray structures.

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